A student was investigating cooling times for two pots made of different materials. One pot was stainless steel and the other pot was iron. The pots were placed on a table in a 25°C room. The pots were roughly the same size and shape, and contained the same amount of water. The pots and water were originally at 100°C. The student recorded the temperature of the water in each pot over time. The graph below shows the results.



- a. Determine the amount of time it took the water in the iron pot to cool from 100°C to 60°C.
- b. Based on the graph, which pot, the iron pot or the stainless steel pot, was a better conductor of thermal energy? Explain your answer.
- c. Identify **two** methods of heat transfer that occurred as the water in the pots cooled, **and** describe how the transfer of heat occurred for **each** method.
- d. Describe when the pots no longer experienced a net loss of thermal energy.