



Massachusetts Comprehensive Assessment System

Chemistry Formula and Constants Sheet

Common Polyatomic Ions

Ion	Ionic Formula
Ammonium	NH_4^+
Carbonate	CO_3^{2-}
Hydroxide	OH^-
Nitrate	NO_3^-
Phosphate	PO_4^{3-}
Sulfate	SO_4^{2-}

Combined Gas Law:
$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

Ideal Gas Law: $PV = nRT$

Absolute Temperature Conversion: $K = ^\circ\text{C} + 273$

Moles of Solute: $M_1 V_1 = M_2 V_2$

Definition of pH: $\text{pH} = -\log[\text{H}_3\text{O}^+] = -\log[\text{H}^+]$

Molar Volume of Ideal Gas at STP: 22.4 L/mol

Ideal Gas Constant: $R = 0.0821 \text{ L} \cdot \text{atm/mol} \cdot \text{K} = 8.31 \text{ L} \cdot \text{kPa/mol} \cdot \text{K}$

Avogadro's Number: 6.02×10^{23} particles/mol

STP: 1 atm (101.3 kPa), 273 K (0°C)

Nuclear Symbols

Name	Symbol
Alpha particle	α or ${}^4_2\text{He}$
Beta particle	β or ${}^0_{-1}e$
Gamma ray	γ
Neutron	1_0n

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Periodic Table of the Elements

Group (Family)

1	1A	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18																																																															
1	Hydrogen	2A	2	3B	4B	5B	6B	7B	8B	9	10	11	12	3A	4A	5A	6A	7A	8A																																																															
2	Li 3 Lithium 6.941	Be 4 Beryllium 9.01218	Na 11 Sodium 22.98977	Mg 12 Magnesium 24.305	Al 13 Aluminum 26.98154	Si 14 Silicon 28.0855	P 15 Phosphorus 30.97376	S 16 Sulfur 32.06	Cl 17 Chlorine 35.453	Ar 18 Argon 39.948	K 19 Potassium 39.0983	Ca 20 Calcium 40.08	Sc 21 Scandium 44.9559	Ti 22 Titanium 47.88	V 23 Vanadium 50.9415	Cr 24 Chromium 51.996	Mn 25 Manganese 54.9380	Fe 26 Iron 55.847	Co 27 Cobalt 58.9332	Ni 28 Nickel 58.69	Cu 29 Copper 63.546	Zn 30 Zinc 65.39	Ga 31 Gallium 69.72	Ge 32 Germanium 72.59	As 33 Arsenic 74.9216	Se 34 Selenium 78.96	Br 35 Bromine 79.904	Kr 36 Krypton 83.80	Rb 37 Rubidium 85.4678	Sr 38 Strontium 87.62	Y 39 Yttrium 88.9059	Zr 40 Zirconium 91.224	Nb 41 Niobium 92.9064	Mo 42 Molybdenum 95.94	Tc 43 Technetium (98)	Ru 44 Ruthenium 101.07	Rh 45 Rhodium 102.906	Pd 46 Palladium 106.42	Ag 47 Silver 107.868	Cd 48 Cadmium 112.41	In 49 Indium 114.82	Sn 50 Tin 118.71	Sb 51 Antimony 121.75	Te 52 Tellurium 127.60	I 53 Iodine 126.905	Xe 54 Xenon 131.29	Cs 55 Cesium 132.905	Ba 56 Barium 137.33	La 57 Lanthanum 138.906	Hf 72 Hafnium 178.49	Ta 73 Tantalum 180.948	W 74 Tungsten 183.85	Re 75 Rhenium 186.207	Os 76 Osmium 190.2	Ir 77 Iridium 192.22	Pt 78 Platinum 195.08	Au 79 Gold 196.967	Hg 80 Mercury 200.59	Tl 81 Thallium 204.383	Pb 82 Lead 207.2	Bi 83 Bismuth 208.980	Po 84 Polonium (209)	At 85 Astatine (210)	Rn 86 Radon (222)	Fr 87 Francium (223)	Ra 88 Radium 226.025	Ac 89 Actinium 227.028	Rf* 104 Rutherfordium (261)	Db 105 Dubnium (262)	Sg 106 Seaborgium (263)	Bh 107 Bohrium (264)	Hs 108 Hassium (265)	Mt 109 Meitnerium (266)	110	111	112	113	114	115	116	117	118
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Mass numbers in parentheses are those of the most stable or most common isotope.

Lanthanide Series

Actinide Series