

Glucose  $(C_6H_{12}O_6)$  is formed in plants by the process of photosynthesis. The net equation for photosynthesis is shown below.

$$6\text{CO}_2 + 6\text{H}_2\text{O} \xrightarrow{\text{sunlight}} \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$$

- a. Calculate the molar mass of glucose. Show your calculations and include units in your answer.
- b. Explain how the amount of carbon dioxide consumed by a plant can be determined from measuring the amount of oxygen released by the plant. Assume excess water is available.
- c. Calculate the amount of glucose, in grams, formed when 100 mol of  $O_2$  is released. Show your calculations and include units in your answer.